

Letter

Structure refinement of AuMg_3 , IrMg_3 and $\text{IrMg}_{2.8}$

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1. Introduction

We have recently redetermined the crystal structure of Mg_3Pt [1]. It has been shown that Mg_3Pt , crystallizing in the acentric space group $P6_3cm$, is isotypic with Cu_3P [2] and that the structure models proposed by Ferro and Rambaldi (Na_3As type, space group $P6_3/mmc$ [3]) and Mansmann (anti- LaF_3 type, space group $P\bar{3}c1$ [4]) are wrong.

A literature search shows that the same problems as for Mg_3Pt are obviously met for the analogous compounds AuMg_3 and IrMg_3 . Schubert and Anderko [5] originally described AuMg_3 as isotypic with Na_3As ($a=4.63$ Å, $c=8.44$ Å, $Z=2$, space group $P6_3/mmc$). In a later publication, van Look and Schubert [6] corrected this structure type assignment and proposed an anti- LaF_3 type structure for AuMg_3 ($a=8.04$ Å, $c=8.46$ Å, $Z=6$, space group $P\bar{3}c1$). This was confirmed by Daams and van Vucht [7]. It should be noted that these researchers have also considered the possibility of a Cu_3P -type structure for AuMg_3 . They rejected it, however, because the R value, resulting from a comparison of observed and calculated X-ray powder intensities, was slightly lower for the $P\bar{3}c1$ model (0.188) than for the $P6_3cm$ model (0.20).

Much less is known for IrMg_3 . Ferro *et al.* [8], who obviously synthesized this compound for the first time, proposed from X-ray powder patterns an Na_3As -type structure for IrMg_3 ($a=4.549$ Å, $c=8.229$ Å, $Z=2$, space group $P6_3/mmc$).

All previous investigations suffer from the lack of single-crystal data. We have succeeded now in the preparation of single crystals of AuMg_3 and IrMg_3 by high pressure high temperature synthesis. In this connection we found another phase with an approximate 1:3 composition in the Ir-Mg system, namely $\text{IrMg}_{2.8}$.

In the present paper we describe the results of a structure refinement for these three compounds, using single-crystal X-ray data.

2. Experimental details

High pressure experiments were carried out in a modified Belt-type apparatus [9], using capsules made from tantalum and starting with stoichiometric mixtures of the elements. A homogeneous, but microcrystalline, sample of AuMg_3 was obtained after reaction at 12 kbar, 1400 °C (30 min) and subsequent quenching to ambient conditions. Small single crystals had been formed after the following treatment: at 12 kbar heating to 1400 °C within 10 min, cooling to 1000 °C within 30 min, to 750 °C within 70 min, subsequent quenching to ambient conditions.

In the Ir-Mg system (typical reaction conditions: 12 kbar, heating to 1400 °C within 10 min, cooling to 800 °C within 100 min, quenching) two hexagonal phases have been found around the 1:3 compositional range which were almost invariably intergrown. They exhibit the same length of the a axis (7.927(1) Å), but differ in the length of their c axes (8.190(5) and 8.560(4) Å). A variation of the reaction conditions did not result in any detectable change in the unit cell parameters. It must therefore be concluded that both phases are separate entities with a definite composition and that no solid solution range exists between them.

3. Structure analysis

After inspection of a number of crystals by Weissenberg and oscillation photographs individuals could be selected which proved to be suitable for a structure determination and were used for data collection on an Enraf-Nonius CAD-4 diffractometer. For the " IrMg_3 " it has been found that for the two intergrown components all reflections with $l \geq 3$ were sufficiently distant to be measured separately. Consequently, reflections with $l=0, 1, 2$ have been excluded from the final refinement. Details of the structure analysis can be taken from Table 1.

The only condition limiting possible reflections found was $l=2n$ for $h\bar{h}0l$, pointing to a glide plane with

TABLE 1. Crystallographic data and details of the crystal structure analysis for AuMg₃, IrMg₃ and IrMg_{2.8}

	AuMg ₃	"IrMg ₃ "	
		Phase A (IrMg ₃)	Phase B (IrMg _{2.8})
Structure type		Mg ₃ Pt (Cu ₃ P)	
Space group		<i>P6₃cm</i>	
<i>a</i> (Å)	8.0594(3)	7.927(1)	7.927(1)
<i>c</i> (Å)	8.4801(5)	8.190(5)	8.560(4)
<i>c/a</i>	1.0522	1.0332	1.0799
<i>V</i> (Å ³)	477.02(7)	445.5(4)	465.9(3)
<i>Z</i>	6	6	6
<i>D_x</i> (g cm ⁻³)	5.64	5.93	5.67
Crystal size (mm ³)	0.07 × 0.04 × 0.04	0.04 × 0.04 × 0.04	
Diffractometer		Enraf-Nonius CAD-4	
Scan mode		ω - 2θ	
Monochromator		Graphite	
Radiation	Cu K α		Mo K α
Range of intensity measurement	2° ≤ θ ≤ 70°		2° ≤ θ ≤ 30°
Intensities measured	1945	2411	2340
Absorption correction		DIFABS [10]	
Correction factors (maximum, minimum)	1.552, 0.665	1.122, 0.861	1.140, 0.826
Unique reflections	193	140	116
<i>R</i> _{int}	0.054	0.057	0.058
Refined parameters	16	16	16
(Δ/σ) _{max}	0.002	0.001	0.001
<i>R</i> , <i>R</i> _w	0.041, 0.039	0.064, 0.031	0.058, 0.031
($\Delta\rho$) _{max, min} (e Å ⁻³)	+2.4, -2.2	+4.2, -4.4	+6.1, -6.5

component *c*/2 in the (11 $\bar{2}$ 0) orientation. Together with the observed Laue class *6/mmm* this result suggested the possible space groups *P6₃/mcm* or *P6₃cm*. By analogy with Mg₃Pt [1], refinement in the acentric space group *P6₃cm* gave immediately very satisfactory results.

After the isotropic refinement a numerical correction for absorption was applied to the original data sets (program DIFABS [10]) before the final refinement was performed (anisotropic for the heavy atoms, isotropic for magnesium; program system SHELX-76 [11]).

In order to obtain some information about the composition of the two components of the "IrMg₃" crystal investigated, we refined the site occupancy factors (SOFs) as well. The SOF for the iridium position was kept constant at 0.500 while all others were allowed to vary. Whereas for component A (with *c* = 8.190 Å) all SOFs refined to their theoretical values for a composition IrMg₃ within one standard deviation, the SOF of Mg(3) for component B (*c* = 8.560 Å) refined to 0.198(41), as compared with the theoretical value of 0.333. All other SOFs for component B are at their theoretical values. Therefore, a composition IrMg_{2.8(3)} can be calculated for component B.

Final residuals were *R* = 0.041, *R*_w = 0.039 for AuMg₃, *R* = 0.064, *R*_w = 0.031 for IrMg₃, and *R* = 0.058, *R*_w = 0.031 for IrMg_{2.8}.

Atomic coordinates and displacement factors for AuMg₃, IrMg₃ and IrMg_{2.8} are given in Table 2 and

TABLE 2. Positional parameters and isotropic displacement factors (angstroms squared) for AuMg₃, IrMg₃ and IrMg_{2.8}

Atom	Site	<i>x/a</i>	<i>y/b</i>	<i>z/c</i>	<i>U</i> _{eq} / <i>U</i> _{iso} ^a
<i>AuMg₃</i>					
Mg(1)	6c	0.2752(18)	0.2752(18)	0.0821(39)	0.003(4)
Mg(2)	6c	0.3754(17)	0.3754(17)	0.4275(36)	0.010(4)
Mg(3)	4b	1/3	2/3	0.2101(31)	0.019(5)
Mg(4)	2a	0	0	0.3248(48)	0.012(7)
Au	6c	0.3288(3)	0.3288(3)	0.75(fixed)	0.004(1)
<i>IrMg₃</i>					
Mg(1)	6c	0.2820(37)	0.2820(37)	0.0695(37)	0.013(5)
Mg(2)	6c	0.3663(28)	0.3663(28)	0.4131(26)	0.006(4)
Mg(3)	4b	1/3	2/3	0.1918(53)	0.043(16)
Mg(4)	2a	0	0	0.2915(54)	-0.003(10)
Ir	6c	0.3274(4)	0.3274(4)	0.75 (fixed)	0.006(1)
<i>IrMg_{2.8}</i>					
Mg(1)	6c	0.2758(37)	0.2758(37)	0.0890(32)	0.016(5)
Mg(2)	6c	0.3605(31)	0.3605(31)	0.4227(19)	0.007(5)
Mg(3) ^b	4b	1/3	2/3	0.2103(65)	0.064(29)
Mg(4)	2a	0	0	0.3069(58)	-0.018(11)
Ir	6c	0.3295(6)	0.3295(6)	0.75 (fixed)	0.007(2)

^a*U*_{eq} for gold and iridium is defined as one-third of the trace of the orthogonalized *U*_{ij} tensor.

^bThe SOF for Mg(3) refined from the theoretical value of 0.333 to 0.20(4), resulting in a composition IrMg_{2.8}.

derived bond distances in Table 3. (Lists of observed and calculated structure factors have been prepared and may be obtained from the authors (K.-J.R.).

TABLE 3. Selected interatomic distances (angstroms) for AuMg₃, IrMg₃ and IrMg_{2.8}

	AuMg ₃	IrMg ₃	IrMg _{2.8}
Au/Ir–Mg(4)	2.725(10)	2.617(5)	2.657(10)
–Mg(3)	2.726(4) (2×)	2.708(9) (2×)	2.680(9) (2×)
–Mg(2)	2.760(31)	2.771(19)	2.813(17)
–Mg(2)	2.819(19)	2.777(22)	2.868(20)
–Mg(1)	2.845(20) (2×)	2.848(26) (2×)	2.791(26) (2×)
–Mg(1)	2.849(34)	2.641(31)	2.933(28)
–Mg(2)	3.229(18) (2×)	3.068(19) (2×)	3.116(19) (2×)
–Mg(1)	3.494(18)	3.431(25)	3.419(25)
Mg(1)–Au/Ir	2.845(20) (2×)	2.848(25) (2×)	2.791(24) (2×)
–Au/Ir	2.849(34)	2.641(31)	2.933(28)
–Mg(2)	3.014(24) (2×)	2.928(29) (2×)	2.954(29) (2×)
–Mg(4)	3.026(37)	2.881(39)	2.874(41)
–Mg(2)	3.039(45)	2.892(37)	2.934(33)
–Mg(4)	3.111(38)	3.190(42)	3.258(45)
–Mg(3)	3.142(19) (2×)	3.037(29) (2×)	3.078(31) (2×)
–Au/Ir	3.494(18)	3.431(25)	3.419(25)
Mg(2)–Au/Ir	2.760(31)	2.771(19)	2.813(17)
–Au/Ir	2.819(19)	2.777(22)	2.868(20)
–Mg(1)	3.014(25) (2×)	2.928(32) (2×)	2.954(33) (2×)
–Mg(1)	3.039(45)	2.892(37)	2.934(33)
–Mg(1)	3.106(24)	3.068(31)	3.215(31)
–Mg(3)	3.134(26) (2×)	3.105(32) (2×)	3.125(38) (2×)
–Mg(4)	3.149(18)	3.069(24)	3.025(26)
–Au/Ir	3.229(18) (2×)	3.068(19) (2×)	3.116(20) (2×)
–Mg(3)	3.487(29) (2×)	3.402(35) (2×)	3.539(43) (2×)
Mg(3)–Au/Ir	2.726(4) (3×)	2.708(9) (3×)	2.680(9) (3×)
–Mg(2)	3.134(26) (3×)	3.105(32) (3×)	3.125(38) (3×)
–Mg(1)	3.142(19) (3×)	3.037(29) (3×)	3.078(31) (3×)
–Mg(2)	3.487(30) (3×)	3.402(36) (3×)	3.539(42) (3×)
Mg(4)–Au/Ir	2.725(10) (3×)	2.617(7) (3×)	2.657(10) (3×)
–Mg(1)	3.026(37) (3×)	2.881(39) (3×)	2.874(41) (3×)
–Mg(1)	3.111(38) (3×)	3.190(42) (3×)	3.258(45) (3×)
–Mg(2)	3.149(18) (3×)	3.069(24) (3×)	3.025(26) (3×)

4. Discussion

The results of the structure refinement show that AuMg₃, IrMg₃ and IrMg_{2.8} are isostructural with Mg₃Pt and hence with Cu₃P as well. The structure models previously proposed for these compounds are wrong.

For a complete discussion of the structure the reader is referred to refs. 1 and 2.

The slightly negative values of the isotropic displacement factors for Mg(4) in IrMg₃ and IrMg_{2.8} could be explained by a partial replacement of magnesium by iridium in the 2a position. From a refinement of the SOFs a composition Mg_{1.9(2)}Ir_{0.1(2)} instead of Mg_{2.0} could indeed be calculated. This deviation from the ideal stoichiometry, however, is less than one standard deviation. We believe, therefore, that the negative displacement factors are due to an insufficient correction for absorption rather than to a replacement of magnesium by iridium in the 2a position.

At present there is no convincing explanation for the surprising finding that two basically isostructural phases exist in the Ir–Mg system near the 1:3 composition, with no solid solution range between them, namely IrMg₃ and IrMg_{2.8}. Further work on intermetallic phases in the Ir–Mg system is in progress.

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